

## Stereochemistry Terms-review

- Absolute configuration—the exact relative position of atoms in one enantiomer around a central point in the molecule. Can be assigned as *R* or *S*.
- Achiral molecule—a compound that contains at least one mirror plane, and therefore is superimposable on its mirror image. The compound cannot rotate plane-polarized light.
- Chiral carbon atom—a  $sp^3$  hybridized carbon atom bonded to four different groups/substituents (aka a tetrahedral stereogenic center or chirality center)
- Chiral compound—a compound that lacks an internal mirror plane
- Conformation—the orientation of atoms in 3-dimensional space of a compound that has free rotation around a bond or bonds (i.e. boat or chair conformation)
- Dextrorotatory (+,*d*)—a compound that rotates plane polarized light in a clockwise matter
- Diastereomers—stereoisomers that are not enantiomers. Note: there must be two tetrahedral stereocenters (or an alkene that can be assigned *cis* or *trans*) in the molecule for it to have a diastereomer.
- Enantiomer—a compound that has a non-superimposable mirror image
- Enantiomeric excess (optical purity)—Percent of one enantiomer in a non-racemic mixture ( $[\text{Major isomer} - \text{Minor isomer}] / [\text{Major isomer} + \text{Minor isomer}] \times 100$ )
- Epimerize—in a reaction where a tetrahedral stereocenter with a single absolute configuration (*R* or *S*) is converted to a 50:50 mixture of *R* and *S* isomers at that particular stereocenter.
- Levorotatory (–,*l*)— a compound that rotates plane polarized light in a counter-clockwise matter
- meso* compound—a compound containing 2 or more individually chiral carbons, but so that the compound has an internal mirror plan. Therefore, the compound is achiral.
- Optically active (not synonymous with chiral)—An unequal mixture of enantiomers. The mixture will rotate plane-polarized light in proportion to the excess amount of the major enantiomer
- Optically pure—A single enantiomer isolated without contamination by the other enantiomer
- Racemate—equal amounts of the *R* and *S* enantiomers, does not rotate plane-polarized light
- Racemize—in a reaction where a single enantiomer is converted to a 50:50 mixture of enantiomers
- Specific rotation—The number of degrees that a particular compound rotates light, taking into consideration the concentration and length of the sample tube, reported as  $[\alpha]_D^{25}$  where 25 is the temperature and D is the wavelength of light (Na, D-line)
- Stereocenter—a point in a molecule (often a tetrahedral carbon atom attached to 4 different groups or a double bond that can have *cis/trans* isomers) where an interchange of groups results in a different stereoisomer
- Stereoisomer—when comparing two compounds, the compounds have the same physical attachment of atoms but a different 3-dimensional arrangement
- Stereogenic carbon centers—a carbon atom attached to four different groups (aka a chiral carbon atom or chirality center)
- $2^n$  rule—maximum number of stereoisomers where  $n$  = number of stereocenters (tetrahedral+cis/trans alkene isomers)